# On the Structure of the Series of Oxides $\mathbf{M}^{\mathbf{I}} \mathrm{UMO}_{\mathbf{4}} \mathrm{O}_{16}{ }^{*}$ 

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A family of oxides $M^{\mathrm{II}} \mathrm{UMO}_{4} \mathrm{O}_{16}\left(M^{\mathrm{II}}=\right.$ $\mathrm{Mg}, \mathrm{Mn}, \mathrm{Cd}, \mathrm{Ca}, \mathrm{Hg}, \mathrm{Sr}, \mathrm{Pb}$ ) was recently described (1). Unit-cell dimensions were obtained from powder-diffraction photographs, and atom coordinates were derived from single-crystal intensity measurements on the calcium compound. The resulting structure type was described as triclinic, space group $P \overline{1}$. It is correctly described as monoclinic, space group $P 2 / n$.

The transformation $a^{\prime}=a+b, b^{\prime}=b$, $c^{\prime}=c$ leads to the monoclinic cell dimensions listed in Table I. For none of the seven compounds do the angles $\alpha^{\prime}$ or $\gamma^{\prime}$ differ significantly from $90^{\circ}$. (The angles $\alpha^{\prime}$ are those given for $\alpha$ in Table I, Ref. (1), and are " 90.00 "' in all seven cases. Only one of the derived values of $\gamma^{\prime}$ differs from $90.00^{\circ}$ by more than $0.01^{\circ}$-that for the Mn compound, at $90.05^{\circ}$. $)^{1}$ The corresponding transformation $x^{\prime}=x, y^{\prime}=y-x, z^{\prime}=z$, when applied to the coordinates in Table III, Ref. (1), lead to atom positions that are consistent with the symmetry of $P 2 / n$

[^0]TABLE I
Compounds $M^{11} \mathrm{UMO}_{4} \mathrm{O}_{16}$ : Revised Cell Dimensions, Space Group P2/n

| $M$ | $a(\AA)$ | $b(\AA)$ | $c(\AA)$ | $\beta\left({ }^{\circ}\right)$ |
| :--- | :--- | :--- | :--- | :--- |
| Mg | 11.393 | 6.505 | 7.920 | 89.80 |
| Mn | 11.415 | 6.557 | 8.010 | 89.93 |
| Cd | 11.432 | 6.625 | 8.150 | 90.37 |
| Ca | 11.447 | 6.651 | 8.236 | 90.44 |
| Hg | 11.431 | 6.645 | 8.248 | 90.50 |
| Sr | 11.476 | 6.762 | 8.474 | 90.85 |
| Pb | 11.454 | 6.811 | 8.554 | 90.85 |

TABLE II
Coordinates of the Calcium Compound, Space Group P2/n

| Atom | $N^{a}$ | $x$ | $y$ | $z$ |
| :--- | :--- | :--- | :--- | :--- |
| U | 2 | 0.25 | 0.7041 | 0.75 |
| Ca | 2 | 0.25 | 0.7039 | 0.25 |
| $\mathrm{Mo}(1,2)$ |  | 0.5962 | -0.1702 | 0.7956 |
| $\mathrm{Mo}(3,4)$ |  | 0.0928 | 0.1687 | 0.7958 |
| $\mathrm{O}(1,3)$ |  | 0.933 | 0.151 | 0.246 |
| $\mathrm{O}(2,16)$ |  | 0.9372 | 0.147 | 0.749 |
| $\mathrm{O}(4,13)$ |  | 0.096 | 0.151 | 0.504 |
| $\mathrm{O}(5,8)$ |  | 0.128 | 0.418 | 0.758 |
| $\mathrm{O}(6,7)$ |  | 0.746 | 0.297 | 0.035 |
| $\mathrm{O}(9,11)$ |  | 0.097 | 0.156 | 0.005 |
| $\mathrm{O}(10)$ | 2 | 0.25 | 0.072 | 0.25 |
| $\mathrm{O}(12,15)$ |  | 0.125 | 0.419 | 0.263 |
| $\mathrm{O}(14)$ | 2 | 0.25 | 0.063 | 0.75 |

[^1]within the reported errors. (The $z$ coordinate of Mo(s), 0.7559 , was obviously misprinted. I have assumed it to be 0.7959, which leads to correct Mo-O distances.)

Since neither the cell dimensions nor the atom coordinates need to be significantly changed to achieve the symmetry of $P 2 / n$, the bond lengths and general description of the structure remain unchanged. The $n$ glide plane of $P 2 / n$ requires the systematic
absence of reflections $h 0 l$ with $(h+l)$ odd. A supplementary table of $F$ 's that would permit confirmation of these absences is apparently not available.

## Reference

1. M. K. Lee and S. Jaulmes, J. Solid State Chem. 67, 364 (1987).

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    ${ }^{1}$ In Ref. (1), the cell dimensions for the Ca compound as given in the body of the text do not agree with those listed in the accompanying Table I and were apparently misprinted.

[^1]:    ${ }^{a}$ Number in cell, if other than 4.

